

16-7 The Electric Field

Many common forces might be referred to as “contact forces,” such as your hands pushing or pulling a cart, or a tennis racket hitting a tennis ball.

In contrast, both the gravitational force and the electrical force act over a distance: there is a force between two objects even when the objects are not touching. The idea of a force *acting at a distance* was a difficult one for early thinkers. Newton himself felt uneasy with this idea when he published his law of universal gravitation. A helpful way to look at the situation uses the idea of the **field**, developed by the British scientist Michael Faraday (1791–1867). In the electrical case, according to Faraday, an *electric field* extends outward from every charge and permeates all of space (Fig. 16-22). If a second charge (call it Q_2) is placed near the first charge, it feels a force exerted by the electric field that is there (say, at point P in Fig. 16-22). The electric field at point P is considered to interact directly with charge Q_2 to produce the force on Q_2 .

We can in principle investigate the electric field surrounding a charge or group of charges by measuring the force on a small positive **test charge**. By a test charge we mean a charge so small that the force it exerts does not significantly alter the distribution of those other charges that create the field. If a tiny positive test charge q is placed at various locations in the vicinity of a single positive charge Q as shown in Fig. 16-23 (points a, b, c), the force exerted on q is as shown. The force at b is less than at a because b’s distance from Q is greater (Coulomb’s law); and the force at c is smaller still. In each case, the force on q is directed radially away from Q . The electric field is defined in terms of the force on such a positive test charge. In particular, the **electric field**, \vec{E} , at any point in space is defined as the force \vec{F} exerted on a tiny positive test charge placed at that point divided by the magnitude of the test charge q :

$$\vec{E} = \frac{\vec{F}}{q} \quad (16-3)$$

More precisely, \vec{E} is defined as the limit of \vec{F}/q as q is taken smaller and smaller, approaching zero. That is, q is so tiny that it exerts essentially no force on the other charges which created the field. From this definition (Eq. 16-3), we see that the electric field at any point in space is a vector whose direction is the direction of the force on a tiny positive test charge at that point, and whose magnitude is the *force per unit charge*. Thus \vec{E} has SI units of newtons per coulomb (N/C).

The reason for defining \vec{E} as \vec{F}/q (with $q \rightarrow 0$) is so that \vec{E} does not depend on the magnitude of the test charge q . This means that \vec{E} describes only the effect of the charges creating the electric field at that point.

The electric field at any point in space can be measured, based on the definition, Eq. 16-3. For simple situations involving one or several point charges, we can calculate \vec{E} . For example, the electric field at a distance r from a single point charge Q would have magnitude

$$E = \frac{F}{q} = \frac{kqQ/r^2}{q}$$

$$E = k \frac{Q}{r^2}; \quad \text{[single point charge] } (16-4a)$$

or, in terms of ϵ_0 as in Eq. 16-2 ($k = 1/4\pi\epsilon_0$):

$$E = \frac{1}{4\pi\epsilon_0} \frac{Q}{r^2}. \quad \text{[single point charge] } (16-4b)$$

Notice that E is independent of the test charge q —that is, E depends only on the charge Q which produces the field, and not on the value of the test charge q . Equations 16-4 are referred to as the electric field form of Coulomb’s law.

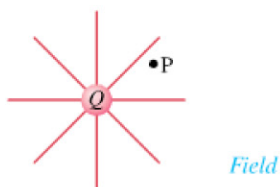
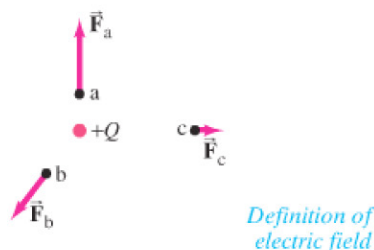


FIGURE 16-22 An electric field surrounds every charge. P is an arbitrary point.

Test charge

FIGURE 16-23 Force exerted by charge $+Q$ on a small test charge, q , placed at points a, b, and c.



Definition of electric field

\vec{E} is a vector

Electric field due to one point charge